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Enantiospecific synthesis of 5-phenylpyrrolo[2,1-c][1,4]benzodiazepines

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article info

ABSTRACT

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Dedicated to Professor Luis Castedo on the occasion of his 70th birthday

Keywords: Pyrrolobenzodiazepine Xanthene Proline Benzodiazepinone Cyclodehydration

The 5-aryl-1,4-benzodiazepin-2-ones constitute an important class of privileged templates as they are able to provide ligands for diverse receptors such as the cholecystokinin receptor (CCK) and several central nervous system (CNS) receptors.^{[1](#page-2-0)} Besides the well-known clinically effective psychoactive drugs such as diazepam[,2](#page-2-0) more than 10,000 benzodiazepines have been found to have pharmacological properties ranging from inhibition of the prolifer-ation of tumor cells,^{[3](#page-2-0)} to antiviral⁴ and analgesic⁵ activities, to the blocking of sodium channel in the treatment of neuropathic pain.^{[6](#page-2-0)}

Pyrrolo[2,1-c][1,4]benzodiazepin-11-ones exhibit different biological properties.⁷ They are useful for treatment of anxiety in warm-blooded animals,⁸ and as a new class of anti-ischemic agents.⁹ Their N₁₀-C₁₁ imino derivatives (PBDs), which can be obtained from them chemically, 10 are gene-specific antitumor agents capable of binding to specific DNA sequences, forming aminal bonds by nucleophilic attack of the $NH₂$ of a guanine at their electrophilic C_{11} C_{11} C_{11} position.¹¹

In view of the biological relevance of the above molecular families, we became interested in the synthesis of hybrids combining the 5-aryl- and pyrrolo-benzodiazepinone frameworks, such as unknown compounds 5-phenylpyrrolo[2,1-c][1,4]benzodiazepin-11 one (1a) and the chromeno-fused derivative 2, a rigid analogue in which the phenyl ring is conformationally frozen by the ether bridge (Fig. 1). A search of the literature showed only two previous reports of this structural motif, both concerning 3,11-diones in

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Enantiomerically pure 5-phenylpyrrolo[2,1-c][1,4]benzodiazepines were synthesized starting from 2-aminobenzophenones and 2-amino-4-methoxyxanthone, using L-proline as a chiral building block.

Figure 1. 5-Phenylpyrrolo[2,1-c][1,4]benzodiazepin-11-ones.

which the pyrrolidinone ring was constructed by intramolecular N-acylation of a 5-phenyl-1,4-benzodiazepin-2-one with a propanoate substituent at position $3¹²$ Furthermore, only two other groups of tricyclic 5-aryl-3,4-fused [1,4]benzodiazepines have been described in the literature, the imidazo $[5,1-c]$ ^{[13](#page-2-0)} and imidazo- $[2,1-c]$ ^{[14](#page-2-0)} derivatives.

For the synthesis of 1a, we started from 2-aminobenzophenone 3a. This was condensed with L-Boc-Pro in the presence of isobutyl chloroformate¹⁵ to provide amide $4a$, which was subsequently deprotected with TFA to afford 5a, both steps proceeding in quantitative yield (Scheme 1). Reduction of the benzophenone carbonyl group with NaBH4 in EtOH at rt gave a 99% yield of benzhydrol 6a as a 10:3 mixture of stereoisomers, as evidenced by 1 H NMR.

All attempts to construct the N_4-C_5 bond of the required diazepinone ring by acidic cyclodehydration failed. Treatment of the diastereomeric mixture **6a** in acetic acid at rt for 48 h had no effect; heating this solution under reflux for 2 h led to extensive decomposition of the starting amide; and heating at 60° C for

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Scheme 1. Synthesis of cis and trans pyrrolo[2,1-c][1,4]benzodiazepines (-)-8 and (-)-9.

24 h afforded (2-acetamidophenyl)(phenyl)methyl acetate as the main product, indicating cleavage of the amide bond of 6a and subsequent N,O-diacetylation of the resulting (2-aminophenyl)(phenyl)methanol in the acetic acid medium. The use of p-TsOH in toluene, whether in catalytic quantities or in large excess, likewise produced no reaction at rt and decomposition at temperatures over 60 \degree C. Stirring for 24 h at rt in methanol saturated with HCl gave a 58% yield of the methyl ether resulting from an intermolecular displacement of the hydroxyl of 6a by the solvent; while treatment with HCl–saturated diethyl ether in dichloromethane for 48 h at rt caused decomposition. Finally, heating at 100 \degree C in PPA for 1 h also led to decomposition of the starting material. Attempts to carry out the cyclodehydration using Lewis acid catalysts such as TiCl $_4$ or BF $_3$ ·OEt $_2$ led to very complex reaction mixtures.

The reluctance of the system to undergo acidic cyclodehydration was attributed to 6a existing mainly as its trans amide bond rotamer, in which the large distance between N_4 and C_5 (4.10 Å according to MM2 calculations) favors alternative reactions of an activated hydroxyl. We inferred that cyclodehydration under neutral conditions would require sufficient thermal energy to bring about rotation to the cis rotamer for the cyclization (Table 1). Nevertheless, prolonged heating of 6a in refluxing toluene achieved no change, regardless of whether water was removed using a Dean– Stark trap (entry 1) or 4 Å molecular sieves (entry 2). Heating in dioxane in a sealed tube at 130 \degree C (entry 3) or in the ionic liquid Bmim¹⁶ at 125 °C (entry 4) was also ineffective, and pre-adsorption on silica followed by heating in an oven at 170 \degree C led to extensive decomposition (entry 5). Finally, however, heating in dichlorobenzene at 180 °C in a sealed tube for 3.5 h (entry 6) led to an excellent 86% yield of the cyclodehydrated product 1a, a yellow solid consisting of a 1:1 mixture of the cis and trans stereoisomers.¹⁷ Chromatography failed to separate both this mixture and its

Table 1 Thermal cyclodehydration conditions tried for amide 6a

Entry	Conditions	Solvent	Time	Temperature	Result
	Dean-Stark	Toluene	24h	Reflux	6a
$\overline{2}$	4ÅMS	Toluene	24h	Reflux	6a
3	4ÅMS	Dioxane	3 _h	130 °C (sealed tube)	6a
4		Bmim	7 h	125 °C	6a
5	SiO ₂ (preabsorbed)		5 min	170 °C	Decomposition
6		DCB	3.5h	180 °C (sealed tube)	1a $(1:1, 86%)$

N-methylated derivative 7a (obtained in 63% yield by treatment of 1a with sodium hydride and methyl iodide in DMF), but LAH reduction of 1a to the corresponding benzodiazepines allowed chromatographic separation of less polar and more polar isomers in 38% and 31% isolated yields, respectively.^{[18](#page-3-0)} Unambiguous identification of the relative stereochemistry of the minor isomer was possible on the basis of a 5% NOE on the intensity of the multiplet at 2.98–3.06 ppm (H_{11a}) upon irradiation of the singlet at 4.86 ppm $(H₅)$, which clearly showed the cis arrangement of these protons (Scheme 1, 8). By contrast, in the major isomer irradiation, of the singlet at 5.19 ppm (H_5) enhanced the multiplets at 2.92– 3.00 ppm $(H_{11\alpha})$ and 2.82–2.87 ppm $(H_{3\alpha})$ by 2.3% and 2.2%, respectively, confirming the location of H_5 on the α -face trans to H_{11a} (Scheme 1, 9).

The optical activity of $(-)$ -8 and $(-)$ -9 suggested preservation of the stereochemistry of C_{11a} (provided by the starting L -proline). Their enantiomeric purity initially seemed to have been confirmed when derivatization of $(-)$ -8 with L-proline afforded what appeared to be a diastereomerically pure amide, 10. Furthermore,

Scheme 2. Synthesis of (14aR,10aS)-pyrrolo[1,2-a]xanthene[1,9-e,f][1,4]diazepin-10-one (2).

when this reaction was carried out on the unseparated mixture of (–)-8 and (–)-9, H_5 singlets at 4.90 and 5.15 ppm in the ¹H NMR spectrum of the product appeared to show the presence of only two stereoisomers, amide 10 and the analogous derivative of $(-)$ -9. However, since the spectrum of the derivatization product of rac-8 (obtained starting from 3 and rac-Pro) did not clearly show amide 10 and the corresponding diastereoisomer, and since derivatization of rac-9 afforded an analogous result, we sought and obtained additional evidence of enantiomeric purity. Successive additions of the chiral shift reagent $Eu(hfc)$ ₃ did not cause chemical shift changes in the ¹H NMR spectrum of (–)-8, but complexation of the carbonyl of 1a by Eu(hfc)₃ shifted its H₅ signals at 4.69 and 5.02 ppm with no splitting, indicating an enantiomeric excess of more than 95%. By contrast, addition of 30% of Eu(hfc)₃ to rac-1a^{[19](#page-3-0)} clearly split its H₅ signals.

We thus accomplished a five-step synthesis of enantiomerically pure 5-phenylpyrrolo[2,1-c][1,4]benzodiazepine 1a from 2-aminobenzophenone in 69% yield. Unfortunately, this protocol seems to be of limited scope, since attempts to cyclize the substituted benzhydrol ${\bf 6b}^{20}$ ${\bf 6b}^{20}$ ${\bf 6b}^{20}$ whether thermally (DCB, 180 °C) or with Lewis acid catalysis (BF₃·OEt₂, DCE, reflux), led to complex mixtures, probably due to the electron-withdrawing halogen substituents making the intermediate carbocation unstable. In the case of the methoxysubstituted benzhydrol $6c₁²¹$ $6c₁²¹$ $6c₁²¹$ heating at 180 °C in DCB led to decomposition, and reaction in acetic acid at rt afforded only in a very modest 18% yield of the cyclodehydration product.

We also synthesized the chromeno-fused pentacyclic derivative 2, a rigid analogue of 1a in which the conformation of the phenyl substituent is fixed by the ether bridge. We started from 4-methoxy-9H-xanthen-9-one (11), which is easily prepared in two steps from commercially available guayacol and o -chlorobenzoic acid.^{[22](#page-3-0)} Treatment of 11 with hot concentrated nitric acid afforded the nitro derivative $\mathbf{12}^{,23}$ $\mathbf{12}^{,23}$ $\mathbf{12}^{,23}$ which was catalytically reduced by hydrogen transfer to give aminoxanthone 13 in 67% overall yield from xanthone 11. Acylation of 13 with N-Boc-L-proline, previously activated with isobutyl chloroformate, provided a quantitative yield of amide 14, which was then deprotected to 15 in 82% yield by treatment with trifluoroacetic acid (Scheme 2). Reduction of 15 with sodium borohydride in refluxing ethanol gave amidoxanthydrol 16, apparently as a mixture of two epimeric alcohols with two rotamers each, the ¹H NMR spectrum showing four singlets for benzylic protons between 5.75 and 5.97 ppm. Finally, reaction of this mixture with AcOH at rt smoothly afforded diazepinone $(+)$ -2 in 70% overall yield from 15.^{[24](#page-3-0)} That this cyclodehydration and that of the open analogue **6c** took place under much milder conditions than those of 6a and 6b suggest that this reaction may proceed through an S_N1 type mechanism, with loss of water from C_5 and the formation of a carbocation intermediate that in the cases of 16 and 6c would be stabilized by the electron-donating substituents.

A distinctive feature of the cyclization of 16 was that it produced just a single stereoisomer, which was identified as the cis derivative 2 by NOE experiments in which irradiation of the singlet at 4.69 ppm (H_{14a}) caused a 1.9% enhancement of the intensity of the doublet at 3.74 ppm (H_{10a}) . This difference in stereoselectivity compared with the reactions of the open analogues **6a** and **6c** can be attributed to the more rigid nature of 16, which leads to the reactions at the two diastereotopic faces of the intermediate xanthylium cation having very different transition states.

In conclusion, we have synthesized enantiopure cis and trans stereoisomers of 5-phenylpyrrolo[2,1-c][1,4]benzodiazepine, $(-)$ -**8**, and $(-)$ -9, in five steps and 69% overall yield from 2-aminobenzophenone, the key step being a non-stereoselective cyclodehydration; and have also stereoselectively prepared the pentacyclic cis-pyrrolo[2,1-c][1,4]benzodiazepinone (+)-2 in six steps from the starting xanthone, in a global yield of 38%.

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¹³C NMR/DEPT (CDCl₃, 75 MHz), δ (ppm): 23.19 and 24.04 (2 × CH₂), 24.73 and 25.51 (2 \times CH₂), 52.17 and 55.71 (2 \times CH₂), 60.84 and 60.86 (2 \times CH), 66.57 and 75.57 (2 \times CH), 121.04 and 122.30 (2 \times CH), 124.38 and 125.45 (2 \times CH), 127.17 and 127.50 (2 \times CH), 127.68 and 127.76 (2 \times CH), 128.17 and 128.42 $(4 \times CH)$, 128.50 and 128.71 ($4 \times CH$), 129.27 and 131.28 ($2 \times CH$), 133.63 (C), 134.96 (C), 135.29 (C), 137.53 (C), 140.21(C), 145.28 (C), 171.94 (CO), 174.49 (CO) . MS (Cl) , m/z $(\%)$: 279 $([M+H]^+, 100)$, 278 $(M^+, 36)$, 251 (20) , 182 (35) . MS (EI), m/z (%): 278 (M⁺, 11), 180 (100). HRMS (EI) calcd for $\mathsf{C}_{18}\mathsf{H}_{18}\mathsf{N}_2$ O: 278.1419, found: 278.1411.
- 18. Reduction of the diastereomeric mixture of lactams $1a$: LiAlH₄ (88 mg, 2.32 mmol) was added portion-wise to a stirred solution of 1a (75 mg, 0.27 mmol) in THF (16 mL), and the resulting mixture was heated under reflux for 6 h. After cooling to 0 °C, the solution was neutralized with HCl (1 M), diluted with ice water, and extracted with dichloromethane $(3 \times 10 \text{ mL})$. The organic extract was washed with brine, dried with $Na₂SO₄$, filtered, and concentrated. Purification by flash chromatography on silica gel (95:5 $CH_2Cl_2/$ MeOH) gave a mixture of the two diastereoisomers of 8 (1:0.8 ratio, 58 mg, 81%), and chromatography of this mixture on silica gel (2:8 hexanes/EtOAc) provided (-)-8 (27 mg, 38%) and (-)-9 (22 mg, 31%) as amorphous solids:

 $(-)$ -(5R,11aS)-5-phenyl-2,3,5,10,11,11a-hexahydro-1H-pyrrolo[2,1-c][1,4]benzodiazepine (**8**): $[\alpha]_D^{20}$ – 7.2 (c 1, CH₂Cl₂). IR (KBr): 3429 (N–H st), 1631 (N–H δ), 1080 (C–N st) cm⁻¹. ¹H NMR (CDCl₃, 300 MHz), δ (ppm): 1.46–1.54 (m, 1H, $H_{1\alpha}$), 1.67–1.78 (m, 2H, $H_{2\alpha}$, H_{2B}), 1.93–2.10 (m, 1H, H_{1B}), 2.34 (q, J = 8.4 Hz, 1H, H_{3B}), 2.77–2.81 (m, 1H, $H_{3\alpha}$), 2.90–2.95 (m, 1H, $H_{11\alpha}$), 2.98–3.06 (m, 1H, H_{11a}), 3.32 (dd, J = 11.9, 2.9 Hz, 1H, H_{11b}), 4.86 (s, 1H, H₅), 6.72–6.76 (m, 3H, ArH), 7.00–7.07 (m, 1H, ArH), 7.25–7.28 (m, 1H, ArH), 7.32–7.35 (m, 2H, ArH), 7.42– 7.43 (m, 2H, ArH). ¹³C NMR/DEPT (CDCl₃, 75 MHz), *δ* (ppm): 21.96 (C₂), 28.77
(C₁), 52.33 (C₁₁), 53.23 (C₃), 65.45 (C_{11a}), 70.61 (C₅), 120.07 (CH), 120.87 (CH),
126.87 (CH), 127.46 (CH), 128.04 (2 × CH), 12 132.78 (C), 142.83 (C), 148.65 (C). MS (CI), m/z (%): 265 ([M+H]⁺, 100), 206 (2), 187 (2). MS (ESI), m/z (%): 265 ([M+H]⁺,100). HRMS (ESI, [M+H]⁺) calcd for $C_{18}H_{21}N_2$: 265.1699, found: 265.1707.

()-(5S,11aS)-5-phenyl-2,3,5,10,11,11a-hexahydro-1H-pyrrolo[2,1-c][1,4]benzodiazepine (**9**): [α] $_{\text{D}}^{20}$ –78.5 (c 1, CH₂Cl₂). ¹H NMR, (CDCl₃, 400 MHz), δ (ppm): 1.49–1.57 (m, 1H, H_{1β}), 1.70–1.87 (m, 2H, H_{2 α}, H_{2β}), 1.90–1.98 (m, 1H, H_{1 α}), 2.82–2.87 (m, 1H, $H_{3\alpha}$), 2.92–3.00 (m, 2H, $H_{11\alpha}$, $H_{11\beta}$), 3.17 (dd, J = 12.9, 2.7 Hz, 1H, $H_{3\beta}$), 3.19–3.25 (m, 1H, H_{11a}), 5.19 (s, 1H, H_5), 6.70 (dd, J = 7.8, 0.8 Hz, 1H, ArH), 6.77 (td, J = 7.4, 1.2 Hz, 1H, ArH), 7.0 (dd, J = 7.5, 1.2 Hz, 1H, ArH), 7.10 (td, J = 7.6, 1.5 Hz, 1H, ArH), 7.20–7.23 (m, 1H, ArH), 7.27–7.30 (m, 2H, ArH), 7.43– 7.45 (m, 2H, ArH). 13C NMR/DEPT, (CDCl3, 100 MHz), d (ppm): 22.52 (C2), 29.04 (C_1) , 51.40 (C_3) , 51.93 (C_{11}) , 57.01 (C_{11a}) , 69.18 (C_5) , 118.87 (CH), 119.86 (CH), 126.65 (CH), 127.71 (CH); 127.77 (2 \times CH), 129.22 (2 \times CH), 129.36 (C), 132.09 (CH), 140.78 (C), 148.41 (C). MS (ESI), m/z (%): 265 ([M+H]⁺, 100). HRMS (ESI $[M+H]^{\dagger}$) calcd for C₁₈H₂₁N₂: 265.1699, found: 265.1707.

- 19. Obtained from D,L-proline and 2-aminobenzophenone as a 1:1 mixture of cis and trans (\pm) -1a.
- 20. Prepared in three steps and 78% yield from commercially available 2-amino-5 chloro-2'-fluorobenzophenone.
- 21. Prepared in three steps and 79% yield from 2-amino-4'-methoxybenzophenone, which was synthesized in 33% overall yield from aniline by conversion to the pivaloylamide, ortho-lithiation and trapping with pmethoxybenzaldehyde, oxidation with $MnO₂$ and hydrolysis.
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- 24. (+)-(14aR,10aS)-6-methoxy-9,10a,11,12,13,14a-hexahydro-10H-pyrrolo[1,2-a]xanthene[1,9-e,f][1,4]diazepin-10-one (2): NaBH₄ (70 mg, 1.85 mmol) was added portion-wise to a stirred solution of amidoxanthone 15 (200 mg, 0.59 mmol) in EtOH (30 mL), and the resulting mixture was heated under reflux for 7 h. After cooling to 0° C, HCl (10%) was added until gas release ceased, and the solvent was then removed under reduced pressure. The residue was partitioned between CH_2Cl_2 and brine, and the organic layer was washed with water, dried with anhydrous Na₂SO₄, filtered, and concentrated, affording 16 as a mixture of stereoisomers. This crude was dissolved in AcOH (10 mL) and stirred at rt for 72 h. After evaporation of the solvent, water was added (10 mL), followed by a few drops of 5 N NaOH to make a slightly basic solution that was extracted with CH_2Cl_2 . The organic extract was washed with brine, dried with Na₂SO₄, filtered, and concentrated. Flash chromatography (SiO₂, 91:9 CH₂Cl₂/MeOH)
gave **2** (127 mg, 70%) as an amorphous solid: [$\alpha_{\rm D}^{\rm 2O}$ +48.4 (c 1, CH₂Cl₂); IR (KBr) 3433 and 3277 (N-H st), 1673 (CO), 1501 (N-H δ) cm⁻¹. ¹H NMR (CDCl₃ 500 MHz), δ (ppm): 1.73–1.79 (m, 2H, H_{11a}, H_{12β}), 2.03–2.05 (m, 1H, H_{12β}), 2.40–2.43 (m, 1H, H_{11a}), 2.62 (q, J = 8.5 Hz, 1H, H_{13b}), 3.15–3.18 (m, 1H, H_{13a}), 3.74 (dd, J = 7.4, 2.1 Hz, 1H, H_{10a}), 3.94 (s, 3H, OCH₃), 4.69 (s, 1H, H_{14a}), 6.74 (d, $J = 8.5$ Hz, 1H, ArH), 6.89 (d, $J = 8.6$ Hz, 1H, ArH) 7.09–7.13 (m, 1H, ArH), 7.27– 7.33 (m, 3H, ArH), 8.28 (s, 1H, NHCO). ¹³C NMR/DEPT (CDCl₃, 125 MHz), δ (ppm): 23.37 (C₁₂), 23.54 (C₁₁), 51.23 (C₁₃), 54.93 (C_{14a}), 56.59 (OCH₃), 60.84 $(C_{10a}$), 111.67 (CH), 111.69 (C), 115.32 (CH), 117.60 (CH), 117.70 (C), 119.55 (C), 123.01(CH), 129.20 (CH), 130.34 (CH), 141.61 (C), 146.01 (C), 151.82 (C), 171.92 (CO). MS (EI), m/z (%): 322 (M+ , 6), 225 (72), 210 (100). HRMS (EI) calcd for C₁₉H₁₈N₂O₃: 322.1317, found: 322.1304.